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Hydroheteroarylation of Alkynes under Mild Nickel Catalysis

Yoshiaki Nakao,* Kyalo Stephen Kanyiva, Shinichi Oda, and Tamejiro Hiyama* Department of Material Chemistry, Graduate School of Engineering, Kyoto University, Kyoto 615-8510, Japan Received April 6, 2006; E-mail: nakao@npc05.kuic.kyoto-u.ac.jp; thiyama@npc05.kuic.kyoto-u.ac.jp

4-Octyne (2a)a

1

1a

1a

1b

1b

1b

ligand

PMe₃

PCyp₃

PMe₃

PCyp₃

PCyp₃

entry

1

2

3

 4^{e}

 $5^{e,f}$

We report herein nickel-catalyzed activation of an Ar-H bond1 of N-protected 3-cyanoindoles followed by addition reaction across alkynes to give 4 (eq 1), demonstrating divergent nickel catalysis² directed by a ligand and/or an N-protecting group. The observed catalysis has been extended to a wide range of heteroarenes and alkynes to give various heteroaryl-substituted ethenes highly chemo-, stereo-, and regioselectively under mild conditions.



As we have recently disclosed,^{2a,b} the reaction of 3-cyano-1methoxycarbonylindole (1a) with 4-octyne (2a) in the presence of Ni/PMe3 catalyst in toluene at 100 °C gave the arylcyanation product 3aa in 68% yield after 18 h (entry 1 of Table 1). isolated a small amount (6%) of (E)-3-cyano-1-methoxycarbonyl-2-(4-octen-4-yl)indole (4aa), which should be derived from the insertion of the alkyne into the Ar-H bond at the C-2 position of 1a. On the other hand, the identical reaction but with tricyclopentylphosphine (PCyp₃) as a ligand instead of PMe₃ gave 4aa in 16% vield together with 3% of 3aa, the activation of the Ar-CN bond being preceded by that of the Ar-H bond (entry 2). With 3-cyano-1-methylindole (1b), the corresponding hydroarylation product 4ba became dominant even under the Ni/PMe₃ catalysis (entry 3),³ and the catalysis was found to be effective even at 35 °C by using PCyp₃ as a ligand to afford **4ba** in 95% yield.⁴ It is worth noting that *the* bond to be activated is controllable by choosing a ligand and/or an N-protecting group to induce a different catalysis of nickel. Use of 1 mol % of the catalyst still worked well to give 4ba in 88% yield, although the reaction took 96 h for completion (entry 5).

A catalytic cycle in Scheme 1 may be suggested that starts with alkyne-coordinating Ni(0) species A,5 which subsequently undergoes oxidative addition of an Ar-H bond at the C-2 position to give alkyne-coordinating Ar-Ni(II)-H intermediate ${\bf B}$, since the reaction proceeds best with heteroarenes having a rather acidic C-H bond (vide infra) adjacent to a heteroatom and an electron-rich nickel species as a catalyst.^{6,7} Hydronickelation⁸ would give aryl-Ni(II)-alkenyl intermediate C, which then would undergo reductive elimination to afford a cis-hydroarylation product 4 and regenerate Ni(0) species.9 A bulky tri(sec-alkyl)phosphine ligand may retard requisite η^2 -coordination of a cyano group¹⁰ and/or the subsequent oxidative addition of the Ar-CN bond in cyano- and alkynecoordinating Ni(0) species D due to a steric reason. Electrondonating substituents on the nitrogen atom would also slow the rate of the oxidative addition of the Ar-CN bond, making the hydroarylation catalysis a predominant pathway.2a,b,10b

The catalysis is found to be effective for a wide range of heteroarenes (Table 2).¹¹ Methyl indole-3-carboxylates (1c-1e) reacted with 2a in modest to good yields, irrespective of a

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PCyp3 (0.010 mmol) were used.

Scheme 1. Plausible Mechanism



Table 1. Nickel-Catalyzed Reactions of 3-Cyanoindoles with

conv. of

1 (%)^b

86

57

69

>95

>95

^a Reactions were carried out using 1 (1.0 mmol), 2a (1.0 mmol), Ni(cod)₂ (0.10 mmol), and a ligand (PMe₃, 0.20 mmol; PCyp₃, 0.10 mmol) in toluene (2.5 mL) at 100 °C. ^{*b*} Based on recovered 1. ^{*c*} Isolated yields. ^{*d*} E:Z = 91:

9. ^e The reaction was carried out at 35 °C. ^fNi(cod)₂ (0.010 mmol) and

vield of

3 (%)^c

68 (**3aa**)

3 (3aa)

<3 (**3ba**)

<3 (3ba)

<3 (3ba)

yield of

4 (%)^c

6 (**4aa**)

16 (4aa)d

35 (4ba)

95 (4ba)

88 (4ba)

time

(h)

18

18

30

18

96

substituent on nitrogen (entries 1-3). Indoles (1f-1h) having an electron-withdrawing functional group at the C-3 position all gave the corresponding hydroarylation products at 35 $^{\circ}$ C (entries 4–6), whereas 1-methyl-3-phenylindole (1i) required 100 °C to react with **2a** (entry 7). The exclusive reaction of the Ar-H bond over the formyl C-H bond of 1g is remarkable in view of the fact that the latter is also activated to undergo addition across an alkyne under Ni/PBu₃ catalysis.^{1b} Substituted imidazoles (entries 8–11) reacted exclusively at the C-2 position as well as benzofuran (1n) and benzothiophene (10) (entries 12 and 13) even in the presence of an Ar-Cl bond (entry 9) or other acidic C-H bonds (entry 11). Benzoxazole (1p) and 4,5-dimethylthiazole (1q) also participated in the reaction (entries 14 and 15). Unsymmetrical internal alkynes, 4-methyl-2-pentyne (2b), and silylacetylenes 2c and 2d all reacted with 1b in excellent regioselectivities to give the adducts having a larger substituent *trans* to the aryl group (entries 16-18).^{12,13}

In conclusion, we have demonstrated a divergent nickel catalysis on Ar-H versus Ar-CN activation and extended the catalysis to hydroheteroarylation¹⁴ of alkynes under mild conditions. The present hydroheteroarylation is applicable to a diverse range of heteroarenes and alkynes to give various heteroaryl-substituted ethenes, versatile synthetic intermediates and/or targets of pharmaceuticals, natural products, and materials, highly chemo-, regio-, and stereoselectively, and thus would be complementary to well-established intermolecular hydroarylation reactions catalyzed by other transition metals.7,9,15 Studies on the detailed mechanism and potential nickel catalysis on other C-H functionalizations^{16,17} will be the subject of our future work.



^{*a*} Unless otherwise noted, all reactions were carried out using a heteroarene (1.0 mmol), an alkyne (1.0 mmol), Ni(cod)₂ (0.10 mmol), and PCyp₃ (0.10 mmol) in toluene (2.5 mL) at 35 °C. ^{*b*} Isolated yields based on the heteroarene. ^{*c*} The reaction was carried out at 50 °C using 4.0 mmol of **2a**. ^{*d*} The reaction was carried out at 100 °C using 3.0 mmol of **2a**. ^{*e*} The reaction was carried out at 50 °C. ^{*f*} The reaction was carried out at 100 °C using 2.0 mmol of **2d**.

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Supporting Information Available: Detailed experimental procedures including spectroscopic and analytical data (PDF). This material is available free of charge via the Internet at http://pubs.acs.org.

References

- For rare examples of stoichiometric and catalytic C-H activation by nickel, see: (a) Kleiman, J. P.; Dubeck, M. J. Am. Chem. Soc. 1963, 85, 1544.
 (b) Tsuda, T.; Kiyoi, T.; Saegusa, T. J. Org. Chem. 1990, 55, 2554. (c) Brunkan, N. M.; Brestensky, D. M.; Jones, W. D. J. Am. Chem. Soc. 2004, 126, 3627. (d) Clement, N. D.; Cavell, K. J.; Jones, C.; Elsevier, C. J. Angew. Chem., Int. Ed. 2004, 43, 1277. (e) Clement, N. D.; Cavell, K. J. Angew. Chem., Int. Ed. 2004, 43, 3845. (f) Ogoshi, S.; Ueta, M.; Oka, M.; Kurosawa, H. Chem. Commun. 2004, 2732. (g) Keen, A. L.; Johnson, S. A. J. Am. Chem. Soc. 2006, 128, 1806.
- (2) For ligand-dependent divergent nickel catalysis on the reactions of benzonitrile with alkynes, see: (a) Nakao, Y.; Oda, S.; Hiyama, T. J. Am. Chem. Soc. 2004, 126, 13904. (b) Nakao, Y.; Oda, S.; Yada, A.; Hiyama, T. Tetrahedron, in press. (c) McCormick, M. M.; Duong, H. A.; Zuo, G.; Louie, J. J. Am. Chem. Soc. 2004, 127, 13904.
- (3) A stereochemically unidentified product derived from the insertion of two molecules of 2a into the Ar-H bond at the C-2 position of 1b was also obtained in 31% yield.
- (4) Other bulky tri(see-alkyl)phosphines, such as PCy₃ and P(*i*-Pr)₃, also gave comparable results (>90% yield estimated by GC), whereas P(*t*-Bu)₃ and PPh₃ were found totally ineffective for the present hydroarylation.
- (5) Preliminary ³¹P NMR studies show that PCyp₃ does not coordinate to the nickel center of Ni(cod)₂ in the absence of an alkyne; see also ref 1f.
- (6) The following experiments could suggest that an electrophilic metalation pathway as a C-H activating step⁷ would be unlikely. (a) The reaction of 3-cyano-2-deuterio-1-methylindole (1b-d₁) with 2a gave the corresponding deuterated adduct 4ba-d₁. (b) The reaction of an equimolar (0.50 mmol) mixture of 1b-d₁ and methyl 1-methylindole-3-carboxylate (1c) with 2a (1.0 mmol) under the identical conditions did not cause intermolecular deuterium crossover. (c) Nickel(II) complexes, such as Ni-(acac)₂ and NiCl₂, failed to catalyze the reaction in the presence or absence of external bases, such as Et₁N and Cs₂CO₃.



- (7) For *trans*-selective hydroarylations of alkynes via electrophilic metalation, see: (a) Jia, C.; Piao, D.; Oyamada, J.; Lu, W.; Kitamura, T.; Fujiwara, Y. *Science* 2000, 287, 1992. (b) Trost, B. M.; Toste, F. D.; Greenman, K. J. Am. Chem. Soc. 2003, 125, 4518. (c) Tunge, J. A.; Foresee, L. N. Organometallics 2005, 24, 6440.
- (8) Similar regiochemistry was also reported for hydronickelation of unsymmetrical alkynes; see entries 16–18 of Table 2 and refs 1b and 1f.
- (9) For representative examples of *cis*-hydroarylation of alkynes via oxidative addition of an Ar-H bond at high temperature or under irradiation, see: (a) Hong, P.; Cho, B.-R.; Yamazaki, H. *Chem. Lett.* **1979**, 339. (b) Tokunaga, Y.; Sakakura, T.; Tanaka, M. *J. Mol. Catal.* **1989**, 56, 305. (c) Aulwurm, U. R.; Melchinger, J. U.; Kisch, H. *Organometallics* **1995**, *14*, 3385. (d) Kakiuchi, F.; Yamamoto, Y.; Chatani, N.; Murai, S. *Chem. Lett.* **1999**, 615. (f) Kakiuchi, F.; Sato, T.; Igi, K.; Chatani, N.; Murai, S. *Chem. Lett.* **1999**, 615. (f) Kakiuchi, F.; Sato, T.; Igi, K.; Chatani, N.; Murai, S. *Chem. Lett.* **2001**, 386. (g) Lim, S.-G.; Lee, J. H.; Moon, C. W.; Hong, J.-B.; Jun, C.-H. *Org. Lett.* **2003**, *5*, 2759. (h) Kuninobu, Y.; Tokunaga, Y.; Kawata, A.; Takai, K. J. Am. Chem. Soc. **2006**, *128*, 202. The mechanism of *cis*-hydroarylation of alkynes catalyzed by dinuclear palladium complexes is unclear. See: (i) Tsukada, N.; Mitsuboshi, T.; Setoguchi, H.; Inoue, Y. J. Am. Chem. Soc. **2003**, *125*, 12102.
- (10) (a) García, J. J.; Jones, W. D. Organometallics 2000, 19, 5544. (b) García, J. J.; Brunkan, N. M.; Jones, W. D. J. Am. Chem. Soc. 2002, 124, 9547.
- (11) At present, the following (hetero)arenes have failed to participate in the reaction: 4-(trifluoromethyl)benzonitrile, methyl 4-(trifluoromethyl)benzoate, ethyl 1-methylindole-2-carboxylate, 1-methylpyrrole, 1-methyl imidazole. Unsubstituted indoles gave a mixture of 2- and 3-alkenylated products.
- (12) A similar reaction of 1g with 2c in the presence of a Ru(H)₂(CO)(PPh₃)₃ catalyst at 115 °C has been reported to give the corresponding adduct as a mixture of stereoisomers in 42% yield; see ref 9f.
- (13) Terminal alkynes failed to give the corresponding adducts due to rapid background oligomerization of alkynes.
- (14) For selected examples of related hydroheteroarylations of unsaturated bonds, see: (a) Hong, P.; Cho, B.-R.; Yamazaki, H. Chem. Lett. 1980, 507. (b) Jordan, R. F.; Taylor, D. F. J. Am. Chem. Soc. 1989, 111, 778. (c) Chatani, N.; Fukuyama, T.; Kakiuchi, F.; Murai, S. J. Am. Chem. Soc. 1996, 118, 493. (d) Lu, W.; Jia, C.; Kitamura, T.; Fujiwara, Y. Org. Lett. 2000, 2, 2927. (e) Tan, K. L.; Bergman, R. G.; Ellman, J. A. J. Am. Chem. Soc. 2002, 124, 13964. (f) Pittard, K. A.; Lee, J. P.; Cundari, T. R.; Gunnoe, T. B.; Petersen, J. L. Organometallics 2004, 23, 5514. (g) Murakami, M.; Hori, S. J. Am. Chem. Soc. 2003, 125, 4720. (h) Ferreira, E. M.; Stoltz, B. M. J. Am. Chem. Soc. 2003, 125, 9578. (i) Youn, S. W.; Pastine, S. J.; Sames, D. Org. Lett. 2004, 6, 581. (j) Liu, C.; Han, X.; Wang, X.; Widenhoefer, R. A. J. Am. Chem. Soc. 2004, 126, 3700. (k) Tan, K. L.; Park, S.; Ellman, J. A.; Bergman, R. G. J. Org. Chem. 2004, 69, 7329. (l) Grimster, N. P.; Gauntlett, C.; Godfrey, C. R. A.; Gaunt, M. J. Angew. Chem., Int. Ed. 2005, 46, 7515.
- (15) For hydroarylation of alkynes via the Friedel-Crafts alkenylations, see: (a) Yamaguchi, M.; Kido, Y.; Hayashi, A.; Hirama, M. Angew. Chem., Int. Ed. 1997, 36, 1313. (b) Tsuchimoto, T.; Maeda, T.; Shirakawa, E.; Kawakami, Y. Chem. Commun. 2000, 1573. (c) Reetz, M. T.; Sommer, K. Eur. J. Org. Chem. 2003, 3485. (d) Shi, Z.; He, S. J. Org. Chem. 2004, 69, 3669.
- (16) For selected recent work on intermolecular transition-metal-catalyzed C-C bond formation of indoles via C-H activation, see: (a) Sawada, T.; Fuerst, D. E.; Wood, J. L. *Tetrahedron Lett.* **2003**, *44*, 4919. (b) Wang, X.; Lane, B. S.; Sames, D. J. Am. Chem. Soc. **2005**, *127*, 4996. (c) Lane, B. S.; Brown, M. A.; Sames, D. J. Am. Chem. Soc. **2005**, *127*, 8050. (d) Deprez, N. R.; Kalyani, D.; Krause, A.; Sanford, M. S. J. Am. Chem. Soc. **2006**, *128*, 4972.
- (17) For selected recent work on intermolecular transition-metal-catalyzed C-C bond formation of other heteroarenes via C-H activation, see: (a) Pivsa-Art, S.; Satoh, T.; Kawamura, Y.; Miura, M.; Nomura, M. Bull. Chem. Soc. Jpn. 1998, 71, 4667. (b) Mori, A.; Sekiguchi, A.; Masui, K.; Shimada, T.; Horie, M.; Osakada, K.; Kawamoto, M.; Ikeda, T. J. Am. Chem. Soc. 2003, 125, 1700. (c) Sezen, B.; Sames, D. Org. Lett. 2003, 5, 3607. (d) Dick, A. R.; Hull, K. L.; Sanford, M. S. J. Am. Chem. Soc, 2004, 126, 2300. (e) Bellina, F.; Cauteruccio, S.; Mannina, L.; Rossi, R.; Viel, S. J. Org. Chem. 2005, 70, 3997. (f) Campeau, L.-C.; Rousseaux, S.; Fagnou, K. J. Am. Chem. Soc. 2005, 127, 18020. (g) Campeau, L.-C.; Parisien, M.; Jean, A.; Fagnou, K. J. Am. Chem. Soc. 2006, 128, 581. (h) Lewis, J. C.; Wu, J. Y.; Bergman, R. G.; Ellman, J. A. Angew. Chem., Int. Ed. 2006, 45, 1589.

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